

AS-2162

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M.Sc. (THIRD SEMESTER) EXAMINATION, 2013

CHEMISTRY

(ELECTIVE SUBJECT)

PAPER : CMT-307

(MEDICINAL CHEMISTRY)

SECTION A

Q1 i) WHAT IS THE BASIC PRINCIPLE OF ANCHOR/LINKER CONCEPT?  
WHAT IS PROTECTING GROUP STRATEGY?

ANS i) ANCHOR/LINKER : IT IS A MOLECULAR UNIT COVALENTLY ATTACHED TO THE POLYMER CHAIN MAKING UP THE SOLID SUPPORT.

DIFFERENT LINKERS ARE USED DEPENDING UPON

- THE FUNCTIONAL GROUP THAT WILL BE PRESENT ON THE STARTING MATERIAL.
- THE FUNCTIONAL GROUP THAT IS DESIRED ON THE FINAL PRODUCT ONCE IT IS RELEASED

PROTECTING GROUPS AND SYNTHETIC STRATEGY: WHEN A MOLECULE IS BEING CONSTRUCTED BY SOLID PHASE SYNTHESIS, IT IS IMPORTANT TO PROTECT ANY REACTIVE FUNCTIONAL GROUPS THAT ARE NOT MEANT TO REACT DURING A REACTION SEQUENCE, TO PREVENT SIDE REACTIONS, THE PROTECTING GROUPS SHOULD BE STABLE TO THE REACTION CONDITIONS BUT BE CAPABLE OF BEING REMOVED IN HIGH YIELD UNDER MILD CONDITIONS ONCE SYNTHESIS IS COMPLETE.

ii) DIFFERENTIATE BETWEEN A NEUROTRANSMITTER, HORMONE AND A DRUG.

ANS ii) NEUROTRANSMITTER AND HORMONE CAN BE DISTINGUISHED

- BY THE ROUTE THEY TRAVEL
- THE WAY THEY ARE RELEASED

\* BUT THEIR ACTION WHEN THEY REACH THE CELL IS THE SAME  
NEUROTRANSMITTER ACT AS A CHEMICAL MESSENGER THAT BIND TO RECEPTORS. EXAMPLE ACETYL CHOLINE, NORDRENALINE.

HORMONES - BLOOD SUPPLY (EPINEPHRINE, GLOUCAGON).

NEUROTRANSMITTERS - NERVES.

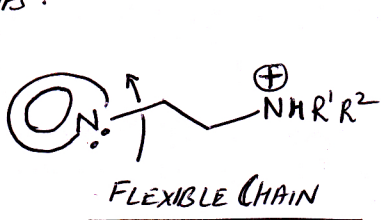
DRUGS - SYNTHETIC ANALOGS THAT WILL BIND TO THE RECEPTOR TO BRING ABOUT ANTAGONIST/AGONIST TYPE RESPONSE.

SECTION A

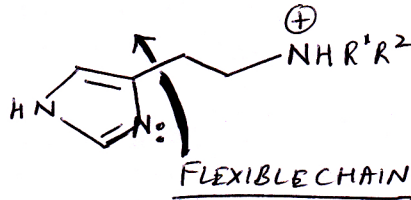
Q1 iii) BASED ON SAR EXPLAIN AGONISTS AT H<sub>1</sub> AND H<sub>2</sub> RECEPTOR?

ANS ii) BASED ON SAR (STRUCTURE ACTIVITY RELATIONSHIP).

AGONISTS AT H<sub>1</sub> & H<sub>2</sub> RECEPTOR NEED TO HAVE THE FOLLOWING STRUCTURAL MOTIFS.



AGONIST AT H<sub>1</sub> RECEPTOR



AGONISTS AT H<sub>2</sub> RECEPTOR.

THE FOLLOWING EXPLAINS THE STRUCTURAL REQUIREMENTS.

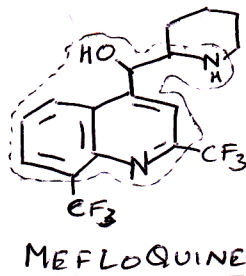
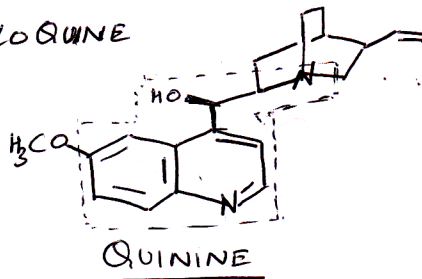
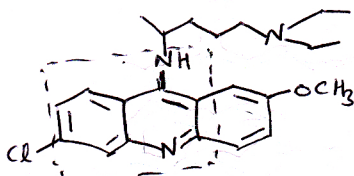
- a) THE SIDE CHAIN HAD TO HAVE A POSITIVELY CHARGED NITROGEN ATOM WITH AT LEAST 1 ATTACHED PROTON FOR EXAMPLE QUATERNARY AMMONIUM SALTS THAT LACKED SUCH A PROTON WERE VERY LOW IN ACTIVITY (AGONIST)
- b) THERE HAD TO BE A FLEXIBLE CHAIN BETWEEN THE ABOVE MENTIONED CATION AND A HETERO AROMATIC RING. (H<sub>1</sub> RECEPTOR)
- c) THE HETERO AROMATIC RING DID NOT HAVE TO BE IMIDAZOLE BUT IT DID HAVE TO CONTAIN A NITROGEN ATOM WITH A LONE PAIR OF ELECTRON; ORTHO TO THE SIDE CHAIN. (H<sub>1</sub> RECEPTOR)

FOR H<sub>2</sub> RECEPTOR SAR REQUIREMENT INCLUDED

- d) THE HETERO AROMATIC RING NEEDED TO CONTAIN AMIDINE UNIT  
(HN-CH-N:)

vii) DRAW THE STRUCTURE FOR QUINACRINE, QUININE AND COMPARE THEIR STRUCTURE TO MEFLOROQUINE

ANS viii)



DOTTED LINES INDICATE THE STRUCTURALLY SIMILAR PARTS OF THE THREE DRUGS.

Q1 ix) HOW IS DICLOFENAC DIFFERENT FROM OTHER NSAIDS?  
[ONLY GIVE KEY POINTS]

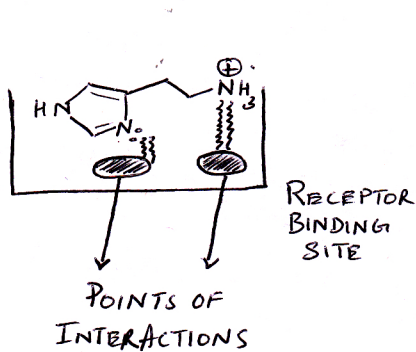
ANS ix) DICLOFENAC SODIUM BELONGS TO THE ARYLALKANOIL ACID CLASS OF ANTI INFLAMMATORY DRUG AND IT DISPLAYS ANTI INFLAMMATORY ANALGESIC AND ANTI PYRETIC PROPERTIES.

THERE ARE THREE MECHANISMS OF ACTION:

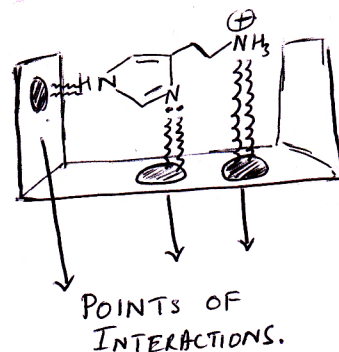
- a) INHIBITION OF ARACHIDONIC ACID COX SYSTEM (WHERE POTENCY IS 3 TO 1000 TIMES MORE POTENT THAN OTHER NSAIDS ON A MOLAR BASIS) RESULTING IN DECREASED PRODUCTION OF PROSTAGLANDINS AND THROMBOXANES
- b) INHIBITION OF THE LIPOXYGENASE PATHWAY RESULTING IN DECREASED PRODUCTION LEUKOTRIENES. PARTICULARLY THE PROINFLAMMATORY  $LKB_4$
- c) INHIBITION OF ARACHIDONIC ACID RELEASE AND STIMULATION OF ITS REUPTAKE, RESULTING IN A REDUCTION OF ARACHIDONIC ACID AVAILABILITY.

x) WHAT ESSENTIAL SUBSTITUTIONS AND INTERACTIONS ARE NECESSARY FOR ANTHISTAMINIC ACTIVITY?

ANS x) CONSIDERING THE FOLLOWING BINDING SCHEMES.



$H_1$  RECEPTOR  
BINDING SITE



$H_2$  RECEPTOR  
BINDING SITE

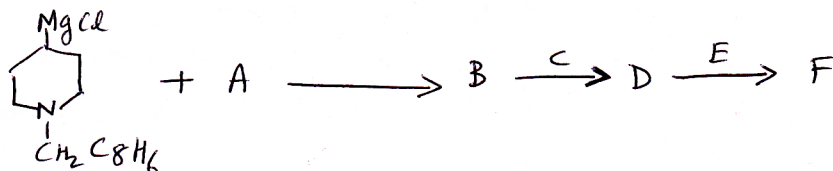
★ ESSENTIAL

THE TWO DIAGRAMS SHOW THE INTERACTIONS NECESSARY FOR ANTIHISTAMMINIC ACTIVITY; THE ESSENTIAL SUBSTITUTIONS ARE

- a) HETEROAROMATIC RING (HN-CH-N)
- b) IMIDAZOLE RING TO CONTAIN A NITROGEN ATOM WITH A LONE PAIR OF ELECTRON, ORTHO TO THE SIDE CHAIN.

SECTION B

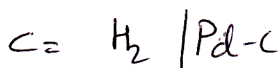
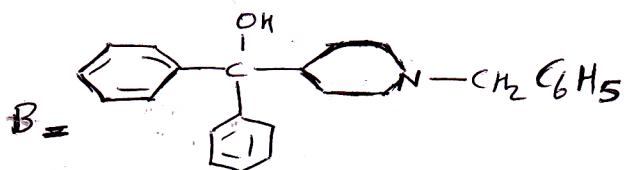
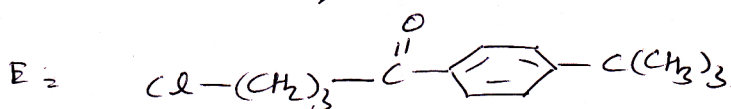
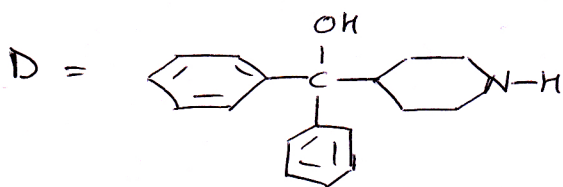
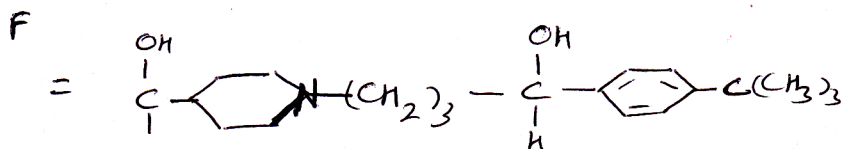
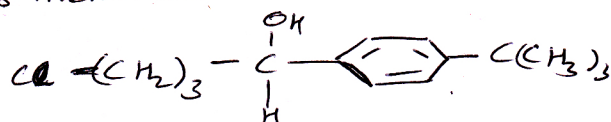
- Q2. a) IDENTIFY THE REACTANTS AND REAGENTS IN THE FOLLOWING SYNTHESIS OF TERFENADINE; EXPLAIN THE SIGNIFICANCE OF REAGENTS IN EACH STEP.



- b) WHAT IS THE PRIMARY MODE OF ACTION FOR TERFENADINE? WHAT ARE THE POINTS OF MODIFICATIONS FOR THIS DRUG TO INCREASE ITS POTENCY?

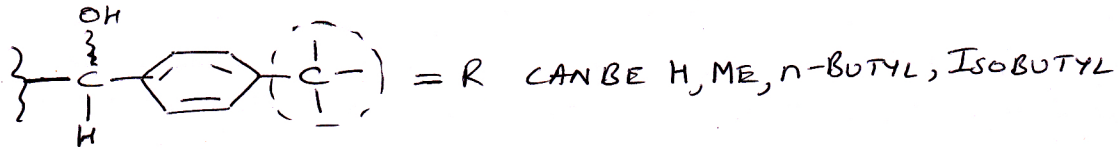
ANS 2

BENZOPHENONE

GRIGNARD REACTION  
NUCLEOPHILIC ADDITIONDEBENZYLACTION, REDUCTION  
DEPROTECTION[ IF THIS REACTANT IS  
USED, WE NEED TO PERFORM  
REDUCTION TO OBTAIN THE  
PRODUCT ]THE CORRECT COMPOUND  
IS THEREFORE

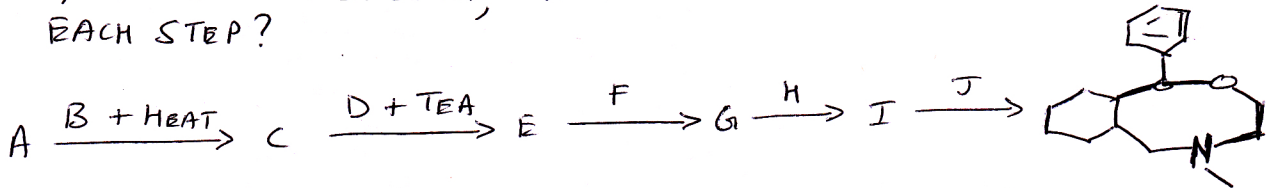
Q2 ANS 2 CONTINUED

b) PRIMARY MODE OF ACTION FOR TERFENADINE; IT COMPETES WITH HISTAMINE FOR BINDING H<sub>1</sub> RECEPTOR SITES IN GI-TRACT UTERUS, LARGE BLOOD VESSELS, BRONCHIAL MUSCLE AND MEDIATES CONTRACTION OF SMOOTH MUSCLES AND INCREASES CAPILLARY PERMEABILITY.



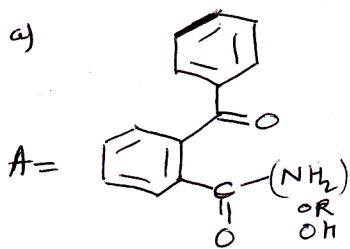
VARIATION IN STRUCTURE PROVIDING ACTIVITY HAS BEEN HIGHLIGHTED ABOVE

Q3 a) IDENTIFY THE REACTANTS AND REAGENTS IN THE FOLLOWING SYNTHESIS OF NEFOPAM. EXPLAIN THE SIGNIFICANCE OF REAGENTS IN EACH STEP?



b) WHAT IS THE PRIMARY MODE OF ACTION FOR NEFOPAM? WHAT ARE THE POINTS OF MODIFICATION FOR THIS DRUG TO INCREASE ITS POTENCY?

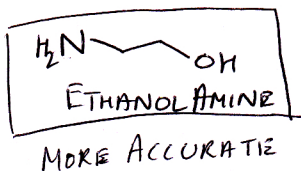
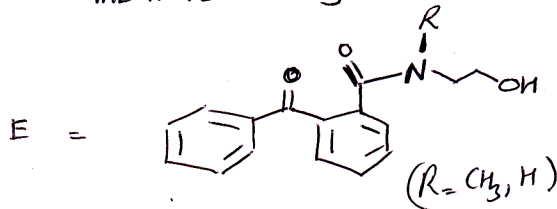
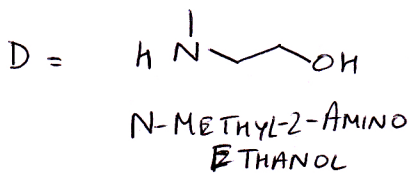
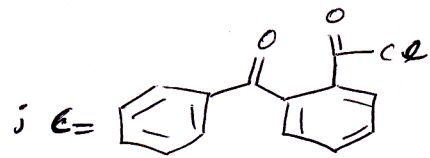
ANS 3 a)



B = SOCl<sub>2</sub>

THIONYL CHLORIDE

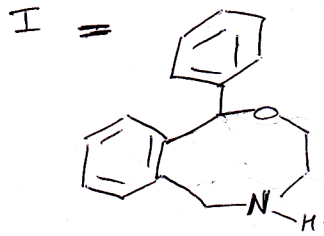
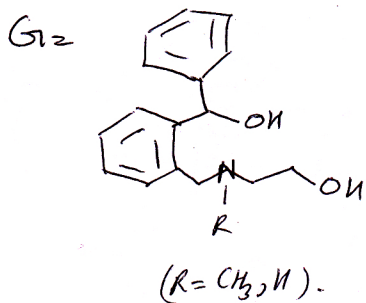
[ACTIVATION OF THE ACYL GROUP]



F = LiAlH<sub>4</sub>

REDUCTION OF KETO TO ALCOHOL

ANS 3 a) CONTINUED



NEFOPAM  
N-METHYLATED

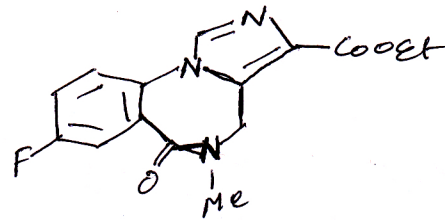
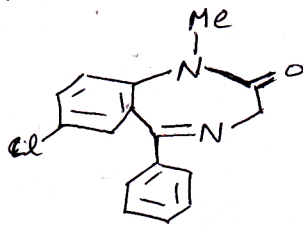
H = p-TSA  
pH, Δ

CYCLIZATION

J = SO<sub>4</sub>Me<sub>2</sub>  
OR CH<sub>2</sub>O

CYCLIZATION  
METHYLATION

b) IT IS A POTENT NON-OPIOID ANALGESIC USED TO IMPROVE POSTOPERATIVE OPIOID ANALGESIA IN PATIENTS. MODIFICATIONS ARE PRIMARILY RELATED TO RING EXPANSION AND SUBSTITUTIONS OF THE FOLLOWING TYPE



SHOWS ENHANCED ACTIVITY

Q 7. WHAT ARE THE ESSENTIAL DIFFERENCES BETWEEN THE BONDING FORCES RESPONSIBLE FOR DRUG BINDING? EXPLAIN THE ROLE OF EACH CLEARLY?

ANS 7. INTERMOLECULAR BONDING FORCES.

A) ELECTROSTATIC OR IONIC BONDS

B) HYDROGEN BONDS (HBD) DONOR (HBA) ACCEPTOR

C) VAN DER WAALS INTERACTIONS

D) DIPOLE-DIPOLE AND ION-DIPOLE INTERACTIONS.

(E) REPULSIVE INTERACTIONS

F) THE ROLE OF WATER AND HYDROPHOBIC INTERACTIONS

EXPLAIN WITH STRUCTURALLY DEFINING THE COMPONENTS IN EACH OF BONDING FORCES.

SECTION A

Q8. WHAT ARE THE ESSENTIAL STEPS INVOLVED IN DRUG DESIGN?

ANS 8. THE FOLLOWING ARE BASIC STRATEGIES.

1. SCREENING OF NATURAL COMPOUNDS FOR BIOLOGICAL ACTIVITY
2. ISOLATION AND PURIFICATION OF THE ACTIVE PRINCIPLE
3. DETERMINATION OF STRUCTURE.
4. STRUCTURE - ACTIVITY RELATIONSHIP (SARS).
5. SYNTHESIS OF ANALOGUES.
6. DESIGN AND SYNTHESIS OF NOVEL DRUG STRUCTURES.

EXPLAIN THE STEPS WITH REFERENCE TO DRUG CLASSIFICATION